

## Color Spectrum of Hydrogen Production and its Future

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### EDITORIAL

The world's energy demand is increasing as per its growing population and standard of living.<sup>[1-2]</sup> The exploitation of fossil fuels to meet these rising energy demands are hence increasing global temperatures due to increased CO<sub>2</sub> and CH<sub>4</sub> concentrations in the atmosphere.<sup>[3-5]</sup> This proposes the urgent requirement to limit the usage of fossil fuels and utilize alternate promising fuels for energy generation.<sup>[6-8]</sup> Hydrogen (H<sub>2</sub>) is a promising alternative clean fuel that can be a game changer in the future of the energy sector. The burning of H<sub>2</sub> only gives energy and water to the exhaust.<sup>[9-10]</sup> Due to its high energy content and clean combustion property, H<sub>2</sub> ensures its place in future energy scenarios<sup>[11]</sup>. H<sub>2</sub> is a colorless gas that various available techniques can produce. The type of H<sub>2</sub> (i.e., grey, green, blue, turquoise, pink, brown, white, etc.) is represented as its color spectrum based on its production technique. The different types of H<sub>2</sub> production techniques via utilization of energy sources such as fossil fuels, hydro, electricity, solar or nuclear, along with their color spectrum, has been discussed in this article.

Green H<sub>2</sub> is referred to as H<sub>2</sub> produced by the electrolysis of water. The energy source utilized to split water into H<sub>2</sub> and O<sub>2</sub> is purely renewable, i.e., solar or wind-powered. This makes it the cleanest way to produce H<sub>2</sub> without leaving carbon footprints.<sup>[12]</sup>

Blue H<sub>2</sub> is a term for the H<sub>2</sub> produced via splitting natural gas into H<sub>2</sub> and CO<sub>2</sub> by employing the reforming processes

such as Steam Reforming of Methane (SRM) or Auto Thermal Reforming of Methane (ATR). However, the  $\text{CO}_2$  produced with  $\text{H}_2$  during the reforming process is captured and stored. Since  $\text{CO}_2$  is the major greenhouse gas trapped in the process without leaving it to the environment, this eliminates the impacts of global warming.<sup>[13]</sup>

Grey  $\text{H}_2$  implies the  $\text{H}_2$  produced by using fossil fuels like natural gas via gas reforming techniques such as SRM or ATR. The process is very similar to that of the blue  $\text{H}_2$  production technique but differs in dealing with  $\text{CO}_2$  produced during the process. Currently, around 95% of the world's  $\text{H}_2$  produced is grey in nature since the  $\text{CO}_2$  produced is not captured and is released free into the atmosphere.<sup>[14]</sup>

Pink  $\text{H}_2$  production is very similar to that of green  $\text{H}_2$ , i.e., by electrolysis of water. The process, however, differs in using the energy source for the process. The power used for pink  $\text{H}_2$  synthesis is nuclear energy, unlike green  $\text{H}_2$ , which uses solar or wind energy.

Another type of  $\text{H}_2$  production process focused rigorously in the present scenario is the Turquoise  $\text{H}_2$ . This process employs the pyrolysis of  $\text{CH}_4$  to produce  $\text{H}_2$  and solid carbon. It is expected that Turquoise  $\text{H}_2$  can be a promising future technique for  $\text{H}_2$  production by using heat produced by renewable energy sources. Furthermore, the produced carbon can be stored in the solid form in a limited space for utilization in several other operations.

The black or brown  $\text{H}_2$  refers to the  $\text{H}_2$  that employs black (bituminous) or brown (lignite) coal via gasification. This type of  $\text{H}_2$  production process is totally opposite of that of green  $\text{H}_2$  and is most damaging to the environment. In addition,  $\text{CO}_2$  and  $\text{CO}$  are produced as a by-product during the process and released directly into the atmosphere. Finally, the white  $\text{H}_2$  refers to the naturally occurring  $\text{H}_2$  in the geological formations underground. These  $\text{H}_2$  deposits are formed by fracking and cannot be used currently due to the deficiency of current exploitation techniques.

For a sustainable future, the usage and dependence on the fossil fuels needs to be reduced. The sustainable development is the only key to fulfill the pledges made during the Paris Agreement and to control the global temperatures is the most urgent amongst them. Therefore, several colors of the  $\text{H}_2$  production spectrum will fade with time while many others will boom. This is due to their simultaneous carbon capture approach along with the clean  $\text{H}_2$  production technique. This fact is supported by recent multibillion-dollar projects signed by different world governments and private ventures, including the USA, KSA, Australia, Germany, U.K, Japan and Russia for commercial blue, green, and turquoise  $\text{H}_2$  projects. Hence, it may be concluded that the future of  $\text{H}_2$  as an energy source lies in the blue, green, and turquoise

spectrum since it is the need of the hour to protect the environment and sustainable development.

## REFERENCES

1. Yusuf M, Farooqi AS, Keong LK, Hellgardt K, Abdullah B. Contemporary trends in composite Ni-based catalysts for CO<sub>2</sub> reforming of methane. Chem Eng Sci. 2021; 229: 116072.
2. Alam MA, et al. Modeling, Optimization and Performance Evaluation of Response Surface Methodology. Materials (Basel). 2021; 14: 4703.
3. Yusuf M, et al. Response surface optimization of syngas production from greenhouse gases via DRM over high performance Ni-W catalyst. Int J Hydrogen Energy. 2021.
4. Yusuf M, et al. Syngas production from greenhouse gases using Ni-W bimetallic catalyst via dry methane reforming: Effect of W addition. Int J Hydrogen Energy. 2021; 46: 27044-27061.
5. Hamid K, et al. Application of polyoxometalate-based composites for sensor systems: A review. J Compos Compd. 2021; 3: 129-139.
6. Rosdin RD, Binti Yusuf M, Abdullah B. Dry reforming of methane over Ni-based catalysts: Effect of ZrO<sub>2</sub> and MgO addition as support. Mater Lett. 2021;X12:100095.
7. Yusuf M, et al. Kinetic studies for DRM over high-performance Ni-W/Al<sub>2</sub>O<sub>3</sub>-MgO catalyst. Int J Hydrogen Energy. 2021;1-10.
8. Bazli L, et al. Application of composite conducting polymers for improving the corrosion behavior of various substrates: A Review. J Compos Compd. 2020; 2: 228-240.
9. Iqbal F, et al. Recent developments in photocatalytic irradiation from CO<sub>2</sub> to methanol. In Nanostructured Photocatalysts- From Fundamental to Practical Applications (eds. Nguyen, V.-H., Vo, D.-V. N. & Nanda, S. B. T.-N. P.). 2021; 519-540.
10. Yusuf M, et al. Syngas Production Employing Ni on Alumina-Magnesia Supported Catalyst Via Dry Methane Reforming. Mater Sci Eng Technol. 2021; 52.
11. Yusuf M, et al. Latest trends in Syngas production employing compound catalysts for methane dry reforming. IOP Conf Ser Mater Sci. 2021; 991: 12071.
12. Yusuf M, Alnarabiji MS, Abdullah B. Clean Hydrogen Production Technologies BT - Advances in

- Sustainable Energy: Policy, Materials and Devices. In: Gao Y, Song W, Liu JL & Bashir S, Editors. 2021; 159-170.
13. Newborough BM, Cooley G, Plc ITMP. Developments in the global hydrogen market : The spectrum of hydrogen colours. Fuel Cells Bull. 2020; 16-22.
  14. Bhandari R, Shah RR. Hydrogen as energy carrier : Techno-economic assessment of decentralized hydrogen production in Germany. Renew Energy. 2021; 177: 915-931.